
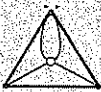
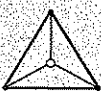


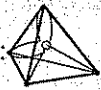
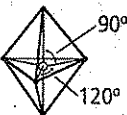
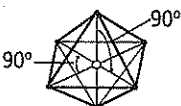


	Molecular shape	Atoms bonded to central atom	Lone pairs of electrons	Type of molecule
Linear		2	0	$AB_2$
Bent or angular		2	1	$AB_2E$
Trigonal-planar		3	0	$AB_3$
Tetrahedral		4	0	$AB_4$
Trigonal-pyramidal		3	1	$AB_3E$
Bent or angular		2	2	$AB_2E_2$
Trigonal-bipyramidal		5	0	$AB_5$
Octahedral		6	0	$AB_6$