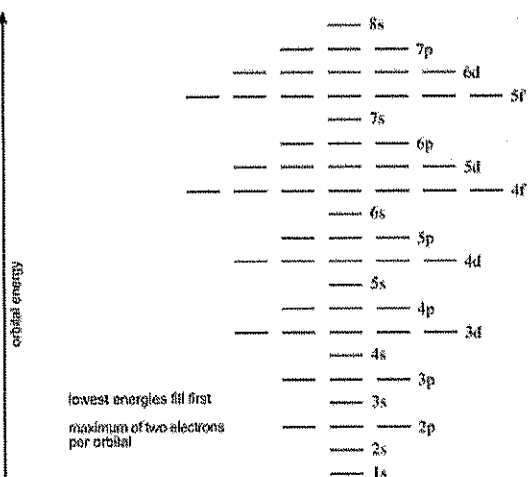
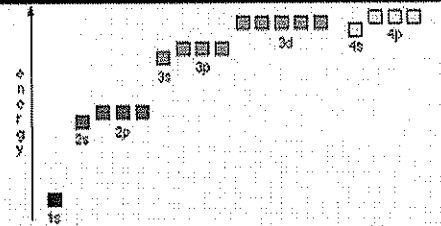


Thus, the orbital filling sequence is, *from the bottom of this diagram, upwards because the lowest energies fill first.*

1s → 2s → 2p → 3s → 3p → 4s → 3d → 4p → 5s → 4d → 5p → 6s → 4f → 5d → 6p → 7s → 5f → 6d → 7p → 8s

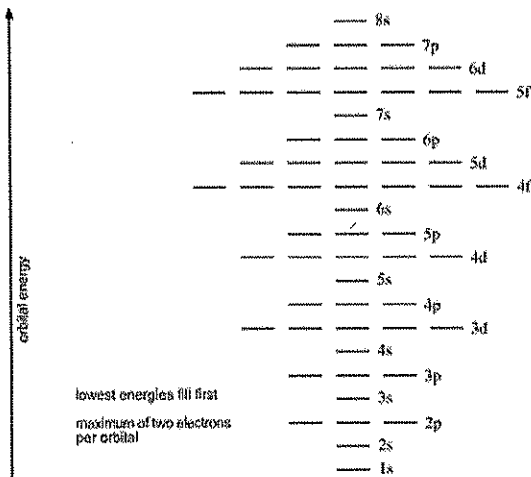


lowest energies fill first
maximum of two electrons per orbital

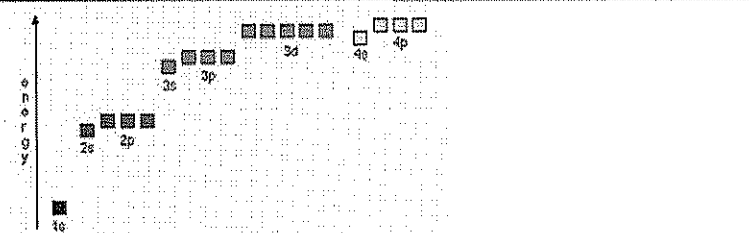


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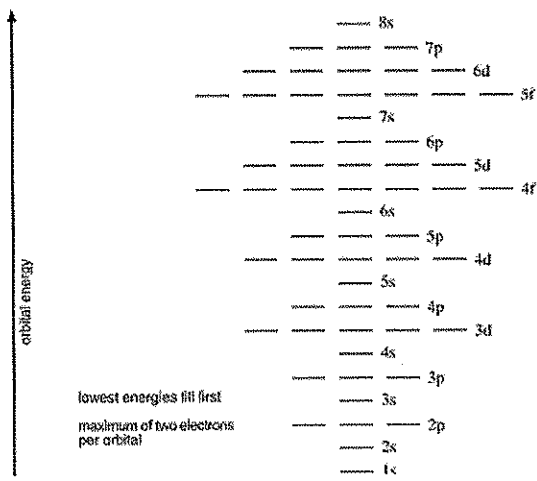


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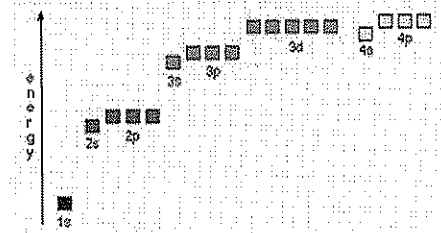


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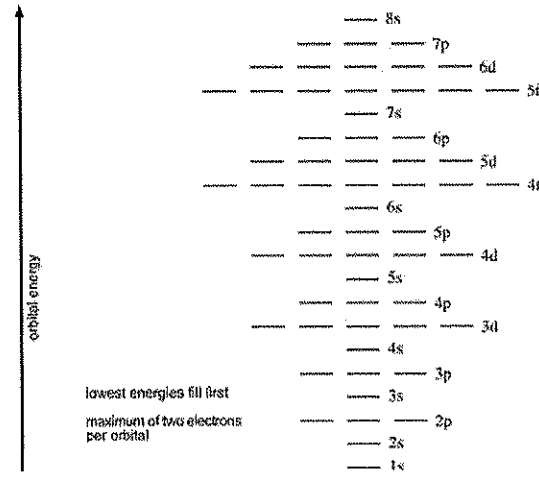


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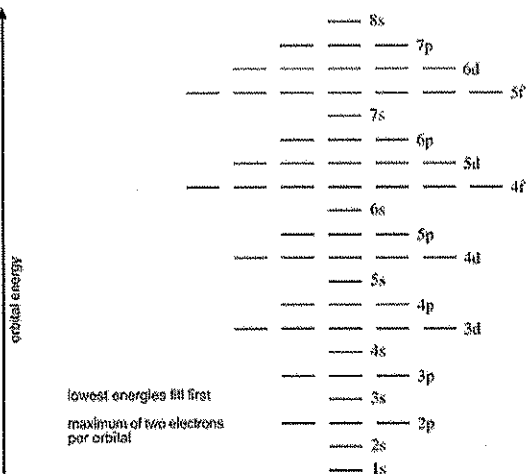
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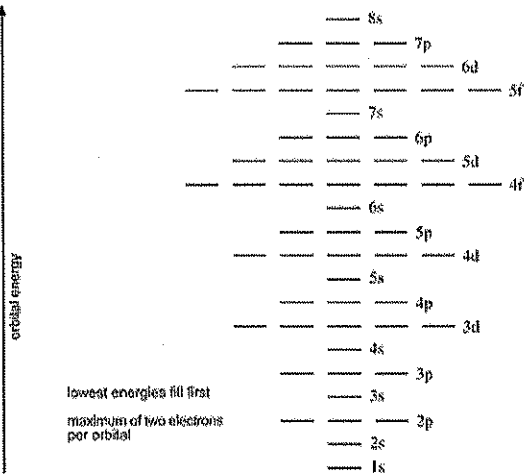
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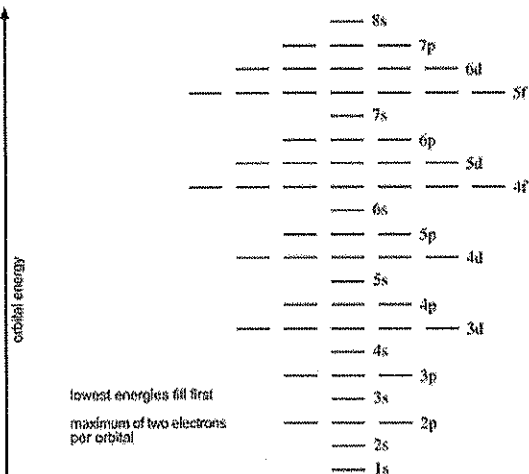
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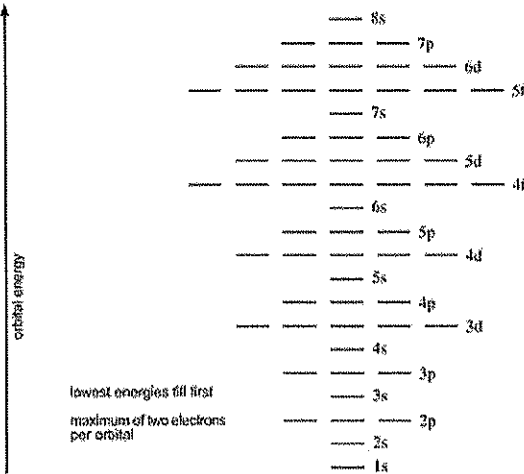
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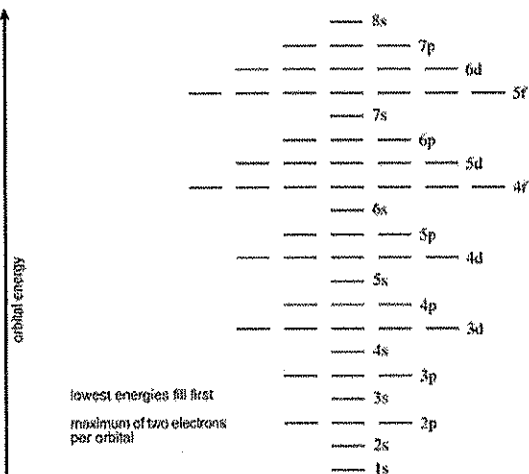
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